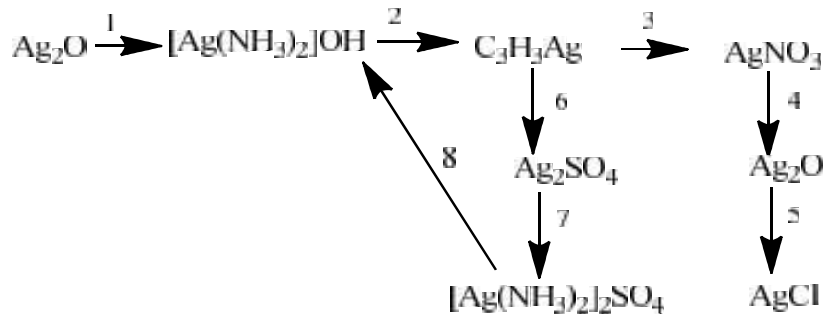


II

1.



20

2.

1. $\text{HClO} + \text{H}_2\text{O}$
2. $+ \text{H}_2\text{O} = 2\text{HClO}$
3. $= 2\text{HClO}_4$
4. $= 3\text{KClO}_3 + \text{KCl}$
5. $= 2\text{KCl} + 3\text{O}_2$
6. $= 2\text{KHSO}_4 + \text{KClO}_4 + 2\text{ClO}_2 + \text{H}_2\text{O}$

- 1.
- 2.
- 3.

1, 4, 6

20

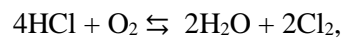
3.

10%- 11,2 NaOH 1,08 / .

- 1.

20

4.



- 1.

16,9.

50%

20

5.

- 1.

- 2.

- 3.

20