

4. $\text{CH}_3\text{CHO} + \text{Cu}(\text{OH})_2 \text{ (t}^0\text{)}$
5. ... ($\text{H}_2\text{SO}_4, \text{t}^0$) $\text{C}_6\text{H}_5\text{NO}_2 + \text{H}_2\text{O}$
6. ... ($\text{Cr}_2\text{O}_3, \text{Al}_2\text{O}_3$) $\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2 + \text{H}_2 + \text{H}_2\text{O}$
7. ... $\text{CH}=\text{CH} + \text{Ca}(\text{OH})_2$
8. ... () $\text{C}_2\text{H}_5\text{OH} + \text{CO}_2$

1. 1 – 4 , , ,
2. 5 – 8 ,
3. , 2.
4. 4 .

20

4. 49 , , 25,3 . 50%

20

5. : , , , , , .
1. , ,
2. ,
3. .

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